defined by the hydrozirconation step. Also, to our knowledge, the direct hydroboration of 1-buten-3-yne to generate dienyls of type **5a** has not been achieved.

These dienyl transfer processes, also referred to generally as transmetalation reactions, operate for a wide variety of metal and metalloid L_nMX species.[9] *One notable exception is Me_3SiCl for which no reaction is observed even at high temperatures.* Attempts to make the silicon center more electrophilic by the use of better leaving groups (e.g. 1-(trimethylsilyl)imidazole or trimethylsilyl trifluoromethanesulfonate) also failed to generate the corresponding 1,3-dienylsilanes. On thermodynamic grounds, it would appear that the Si–X bond energy is too similar to the corresponding Zr–X bond energy[10] and thus there is no strong driving force for transfer from Zr to Si.[11]

These highly functionalized 1,3-dienes are now being evaluated in the Diels–Alder reaction, particularly with respect to stereospecificity and regiochemistry. This and the photolability of these new dienes will be reported in due course.

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Supplementary Material Available: Experimental details, analytical and spectroscopic data for compounds **4a–d**, **4′a–d**, and **5a–d**, and ^{1}H NMR spectra of 4″**b** and 4″**d** (8 pages). Ordering information is given on any current masthead page.

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[9] We have observed similar transfer reactions from Zr to S, Ge, Ga.

[10] The average bond dissociation energy[10b] for Zr–Cl ($ZrCl_4$) is 117 kcal mol^{-1} and the Si–Cl bond dissociation energy in Me_3SiCl is calculated[10c] to be 97.4 kcal mol^{-1}. (b) Huheey, J. E. *Inorganic Chemistry: Principles of Structure and Reactivity*, 3rd ed.; Harper and Row: New York, 1983; p 846. (c) Pilcher, G. Thermochemistry of Organometallic Compounds Containing Metal–Carbon Linkages. *International Review of Science, Physical Chemistry, Series 2*; Butterworths: London, 1975; Vol. 10, Chapter 2, p 45.

[11] A similar lack of reactivity in the transfer reaction from Zr to Si was observed even with use of $SiCl_4$, see ref 6.

Additions and Corrections

Vol. 53, 1988

Giovanni Palmisano,* Bruno Danieli, Giordano Lesma,* Federica Trupiano, and Tullio Pilati. Oxidation of β-Anilinoacrylate Alkaloids Vincadifformine and Tabersonine by Fremy's Salt. A Mechanistic Insight into the Rearrangement of *Aspidosperma* to *Hunteria* Alkaloids.

Page 1057, ref 1. We regret having overlooked the following references to relevant synthesis of *Aspidosperma* alkaloids vindorosine and vindoline when writing our paper and are grateful to Dr. Y. Langlois for calling them to our attention: Andriamialisoa, R. Z.; Langlois, N.; Langlois, Y. *J. Org. Chem.* **1985**, *50*, 961. Gênin, D.; Andriamialisoa, R. Z.; Langlois, N.; Langlois, Y. *Heterocycles* **1987**, *26*, 377.